

HW1

EECS 277B

Solutions

- 1) Calculate the density of states vs. E in a one dimensional world
- 2) Find the spacing between the lowest two energy levels in a 3d box of size 1 m.
- 3) Consider a single electron in a box. Calculate the size of the box at which only the ground state is occupied at room temperature.

$$\begin{aligned} 2) \Delta E &= \frac{\hbar^2}{2m} \left(\frac{\pi}{L}\right)^2 (1^2 + 1^2 + 2^2) - (1^2 + 1^2 + 1^2) \\ &= \frac{3\hbar^2}{2m} \left(\frac{\pi}{L}\right)^2 \end{aligned}$$

$$m = 9.1 \times 10^{-31} \text{ kg}$$

$$\hbar = 1.05 \times 10^{-34} \text{ J-s}$$

$$\begin{aligned} \Delta E &= 1.8 \times 10^{-37} \text{ J} \\ &= 10^{-18} \text{ eV} \end{aligned}$$

$$3) \Delta E > kT \Leftrightarrow \frac{3\hbar^2}{2m} \left(\frac{\pi}{L}\right)^2 > kT$$

$$\Leftrightarrow L < \sqrt{\frac{3\hbar^2}{2m} \frac{\pi^2}{kT}}$$

From above $\Delta E = 10^{-18} \text{ eV}$ (a) $L = 1 \text{ m}$

$$\text{So (a) } L = 1 \text{ m } \frac{\Delta E}{kT} = \frac{10^{-18} \text{ eV}}{\frac{1}{30} \text{ eV}} = 3.3 \times 10^{-17}$$

$$\Rightarrow L = \left[\frac{3.3 \times 10^{-17}}{3.3 \times 10^{-17}} \right]^{\frac{1}{2}} \text{ m} = 5.7 \text{ nm}$$

$$\Rightarrow \text{Need } \underline{L < 5.7 \text{ nm}}$$

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1 dimension

$$N_k dk = ?$$

“Volume” of line segment
=dk

Number of states in length=
length x States/length

$$\text{States/length} = 1 / (\pi/L):$$

$$N_k dk = (dk) \cdot \left(\frac{1}{(\pi / L)} \right) \cdot 2 = \frac{2}{\pi} L dk$$

$$\rho_k dk \equiv \frac{N_k dk}{\text{length}} = \frac{2}{\pi} dk$$

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1 dimension

$$\rho(E)dE = ?$$

We use:

$$\rho_k dk = \rho(E)dE$$

$$\rho_k dk = \frac{2}{\pi} dk$$

$$E = \frac{\hbar^2 k^2}{2m} \Rightarrow k = \sqrt{\frac{2mE}{\hbar^2}} \Rightarrow dk = \sqrt{\frac{2m}{\hbar^2}} \frac{dE}{2\sqrt{E}}$$

$$\rho(E)dE = \frac{2}{\pi} \sqrt{\frac{2m}{\hbar^2}} \frac{dE}{2\sqrt{E}} = \frac{1}{\hbar\pi} \sqrt{\frac{2m}{E}} dE$$