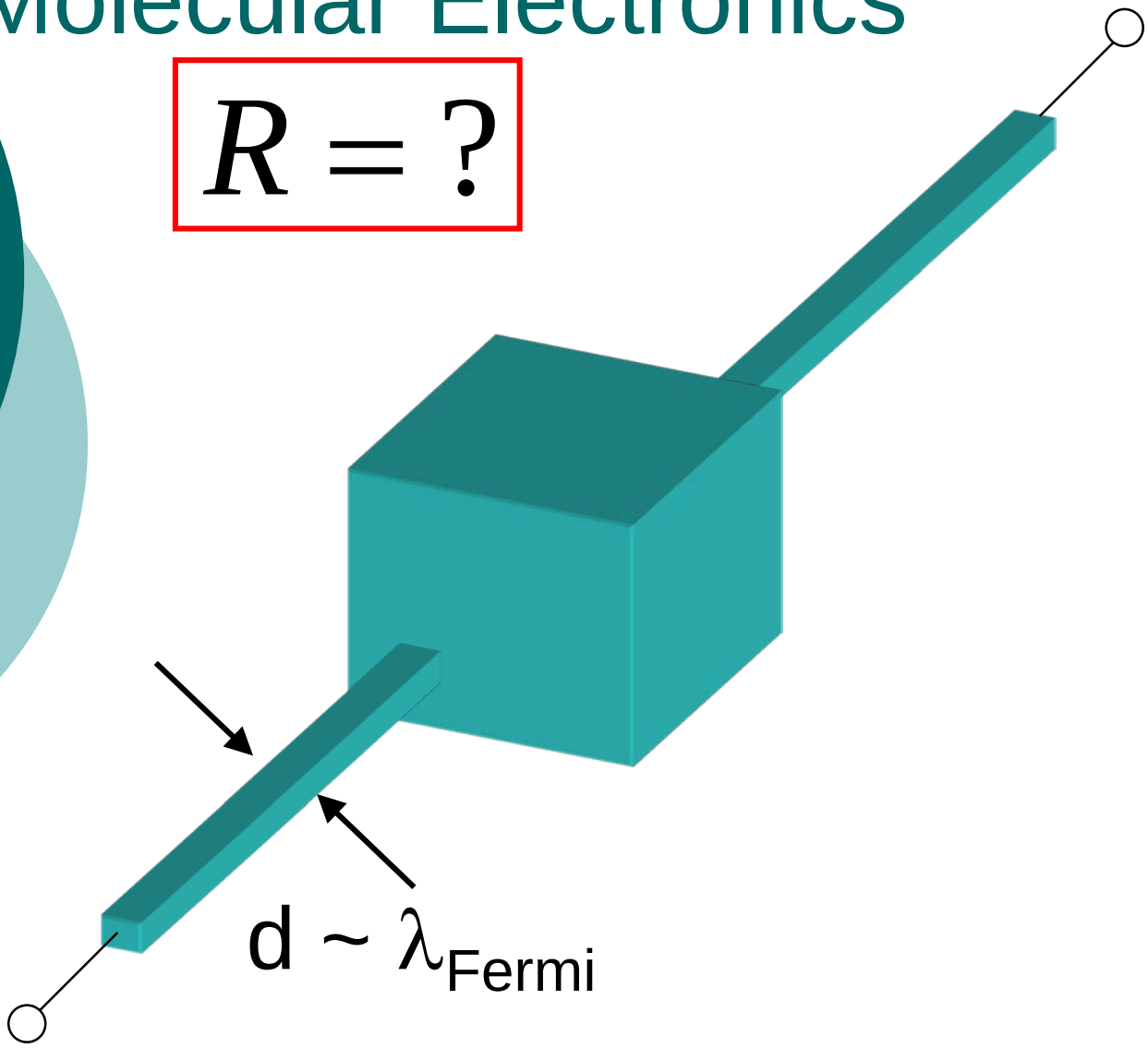
Lecture 12: Quantum dots and Molecular Electronics

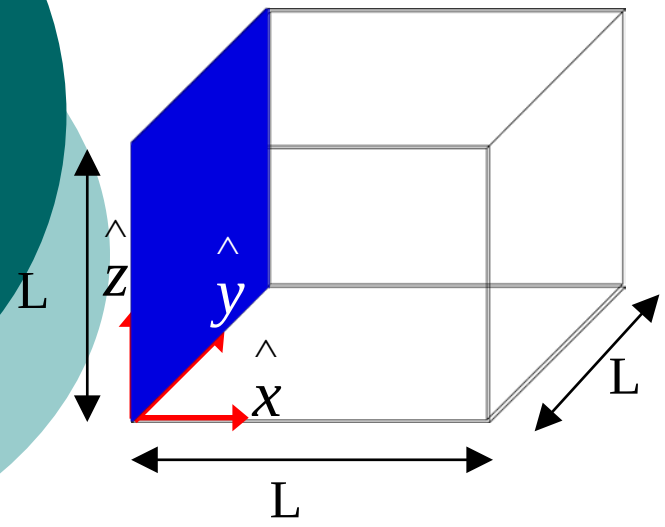
$$R = ?$$



Readings that cover this lecture

- Ferry, pp. 209-226
- Hanson, pp. 125-127

Particle in a box



We can do the same for y, z:

$$\psi(\vec{r}) = (2i)^3 A \cdot \sin(k_{n_x} x) \cdot \sin(k_{n_y} y) \cdot \sin(k_{n_z} z)$$

$$k_{n_x} = n_x \pi / L$$

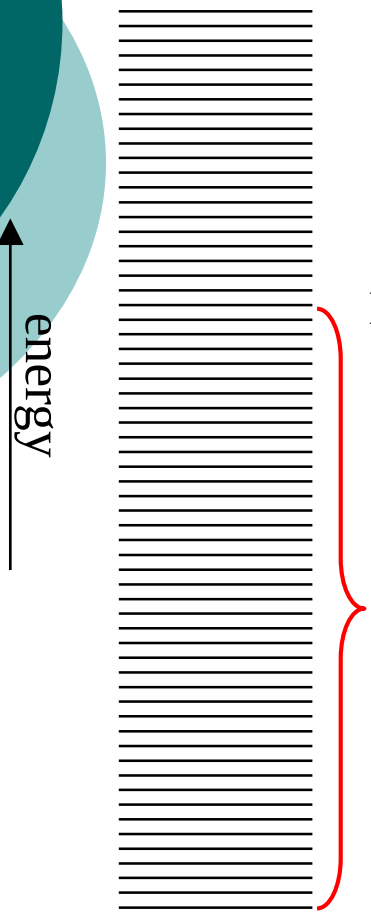
$$k_{n_y} = n_y \pi / L$$

$$k_{n_z} = n_z \pi / L$$

$$E = \frac{\hbar^2 (k_{n_x}{}^2 + k_{n_y}{}^2 + k_{n_z}{}^2)}{2m} = \frac{\hbar^2 (\pi / L)^2}{2m} (n_x{}^2 + n_y{}^2 + n_z{}^2)$$

These are the allowed energy levels, or “quantum states”

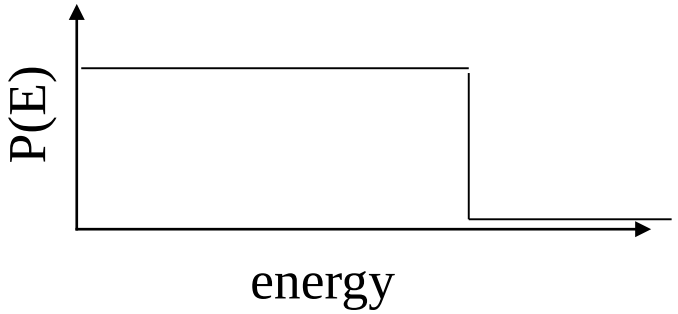
Fermi gas



At zero temperature, as we add electrons to the box, we gradually fill up all the states. (DISCUSS PAULI EXCLUSION PRINCIPLE -IMPORTANT!)

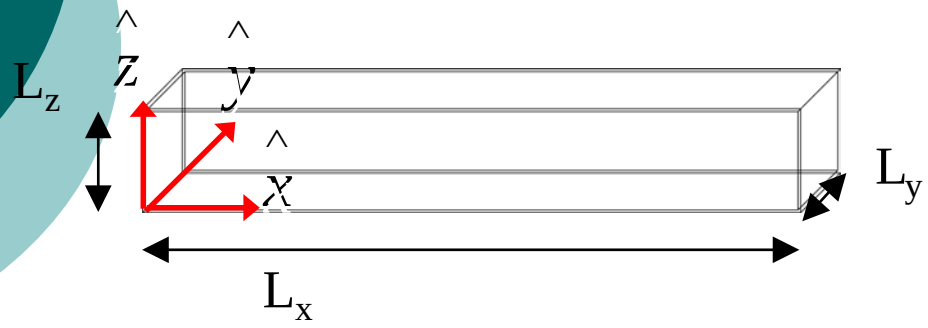
When we are done filling the box, the energy of the last electron is called the "Fermi energy."

"Gas" means we neglect electron-electron interactions.



Particle in a box

$$\psi(\vec{r}) = (2i)^3 A \cdot \sin(k_{n_x} x) \cdot \sin(k_{n_y} y) \cdot \sin(k_{n_z} z)$$



$$k_{n_x} = n_x \pi / L$$

$$k_{n_y} = n_y \pi / L$$

$$k_{n_z} = n_z \pi / L$$

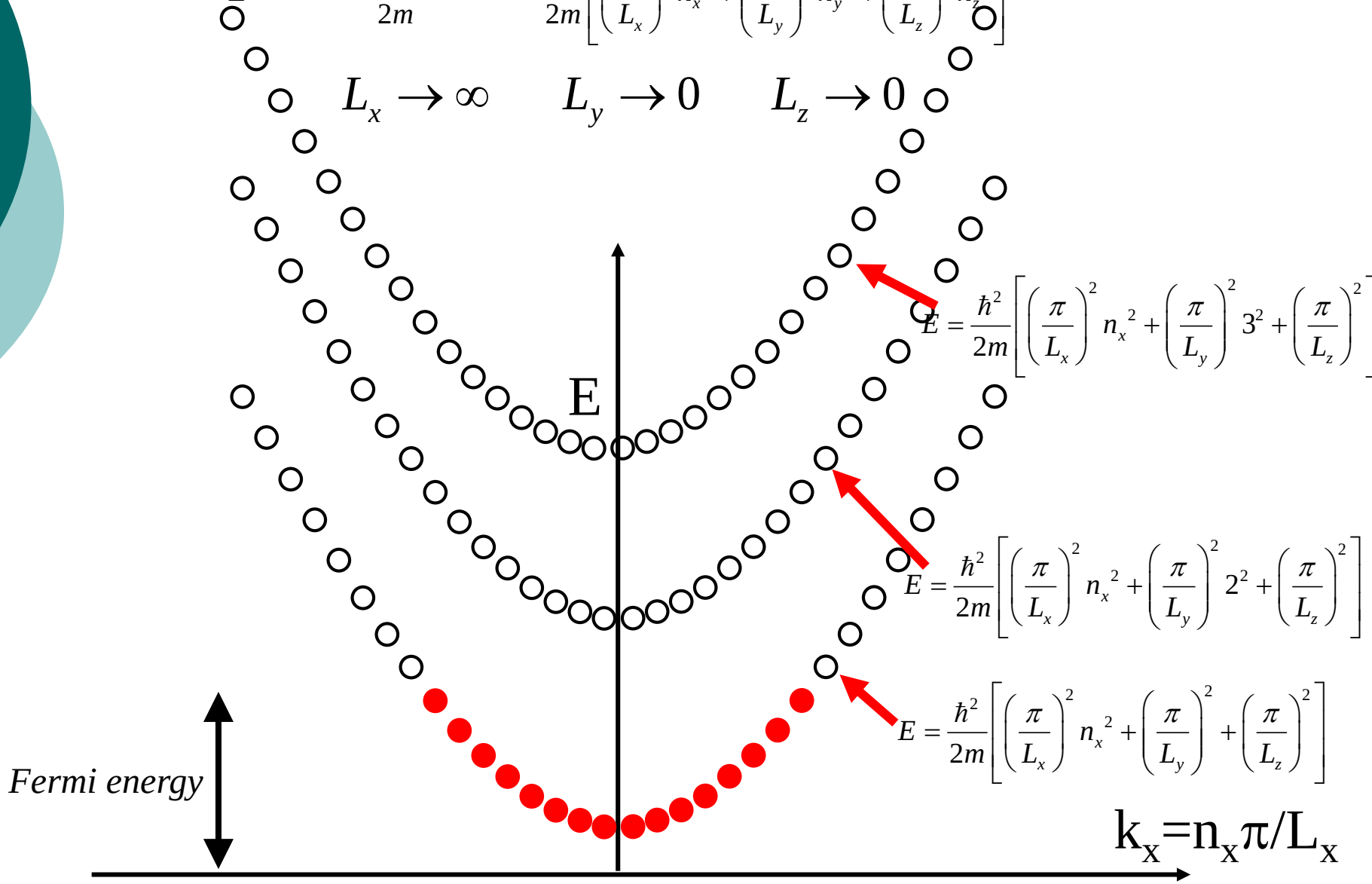
$$E = \frac{\hbar^2 (k_{n_x}{}^2 + k_{n_y}{}^2 + k_{n_z}{}^2)}{2m} = \frac{\hbar^2}{2m}\left[\left(\frac{\pi}{L_x}\right)^2 n_x{}^2 + \left(\frac{\pi}{L_y}\right)^2 n_y{}^2 + \left(\frac{\pi}{L_z}\right)^2 n_z{}^2\right]$$

These are the allowed energy levels, or “quantum states”

1d system:

$$E = \frac{\hbar^2(k_{n_x}{}^2 + k_{n_y}{}^2 + k_{n_z}{}^2)}{2m} = \frac{\hbar^2}{2m}\left[\left(\frac{\pi}{L_x}\right)^2 n_x{}^2 + \left(\frac{\pi}{L_y}\right)^2 n_y{}^2 + \left(\frac{\pi}{L_z}\right)^2 n_z{}^2\right]$$

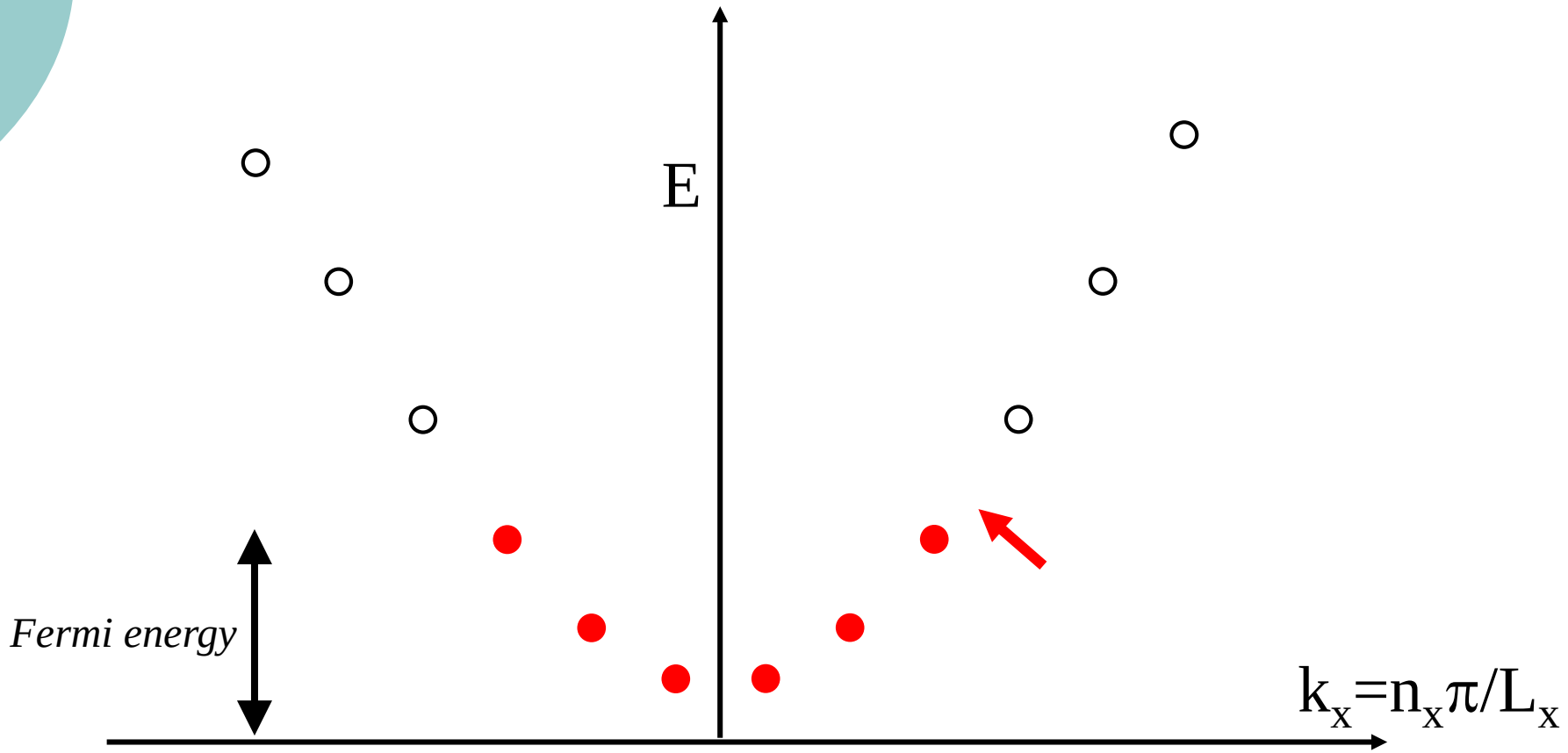
$L_x \to \infty \qquad L_y \to 0 \qquad L_z \to 0$



0d system

$$E = \frac{\hbar^2(k_{n_x}{}^2 + k_{n_y}{}^2 + k_{n_z}{}^2)}{2m} = \frac{\hbar^2}{2m}\left[\left(\frac{\pi}{L_x}\right)^2 n_x{}^2 + \left(\frac{\pi}{L_y}\right)^2 n_y{}^2 + \left(\frac{\pi}{L_z}\right)^2 n_z{}^2\right]$$

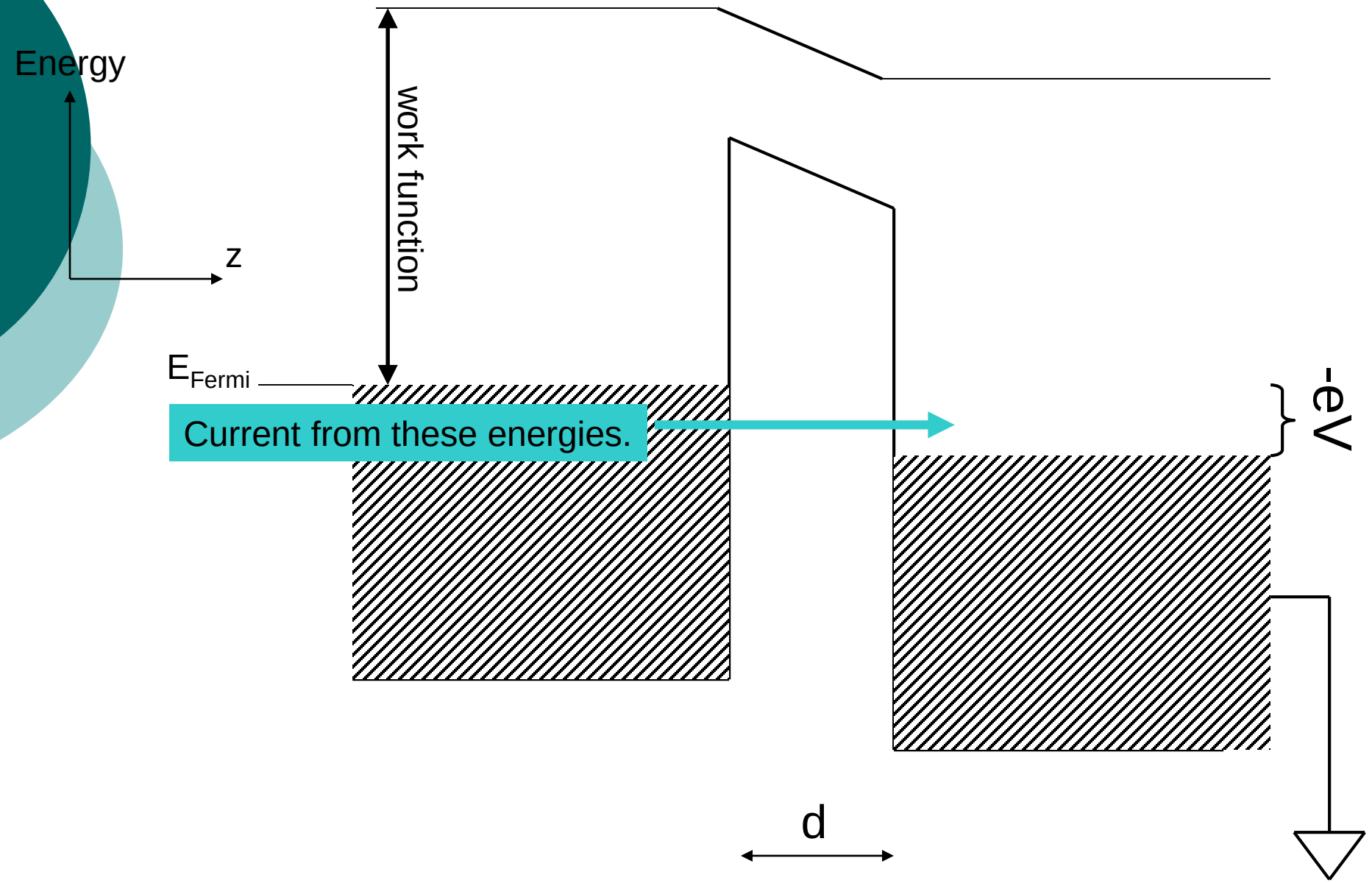
$$L_x \to 0 \qquad L_y \to 0 \qquad L_z \to 0$$



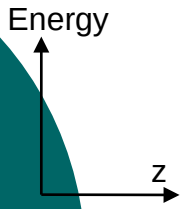
Energy scales

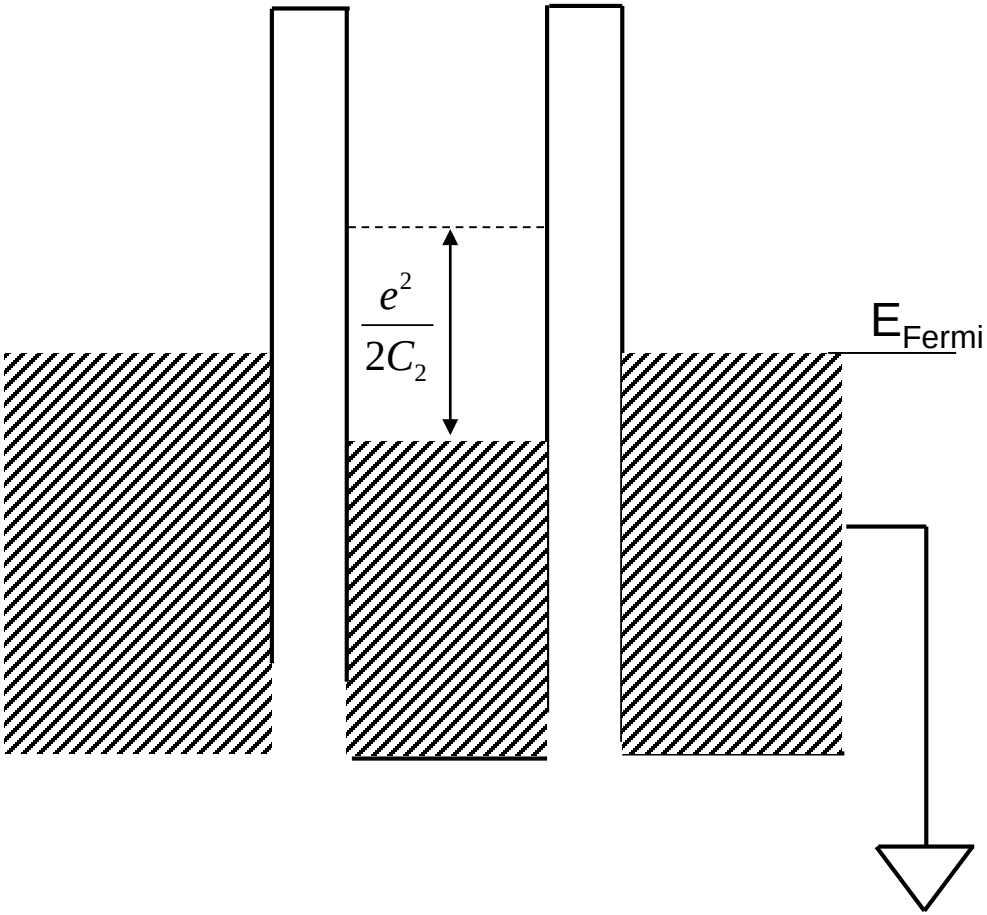
- Charging energy
- Single electron energy level spacing
- Temperature

Continuum band diagrams like this no longer apply

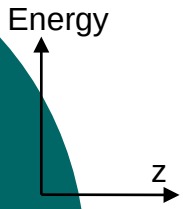


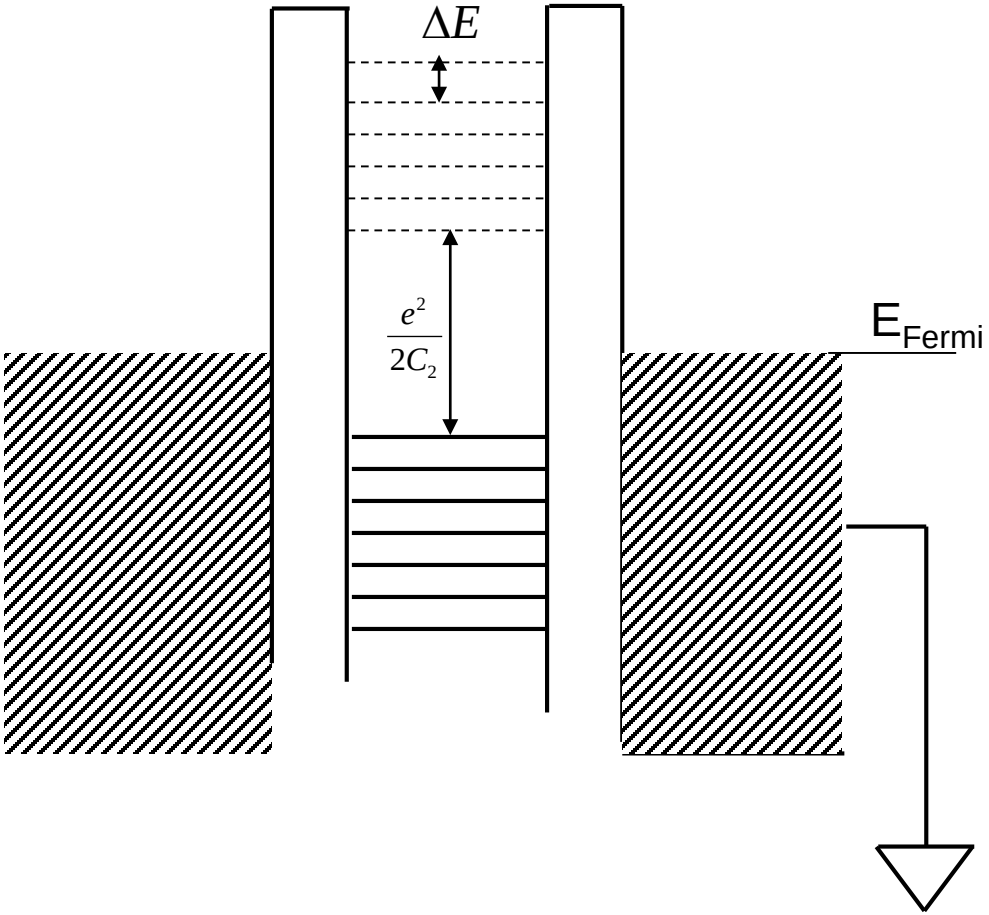
Band diagram with Coulomb "gap"



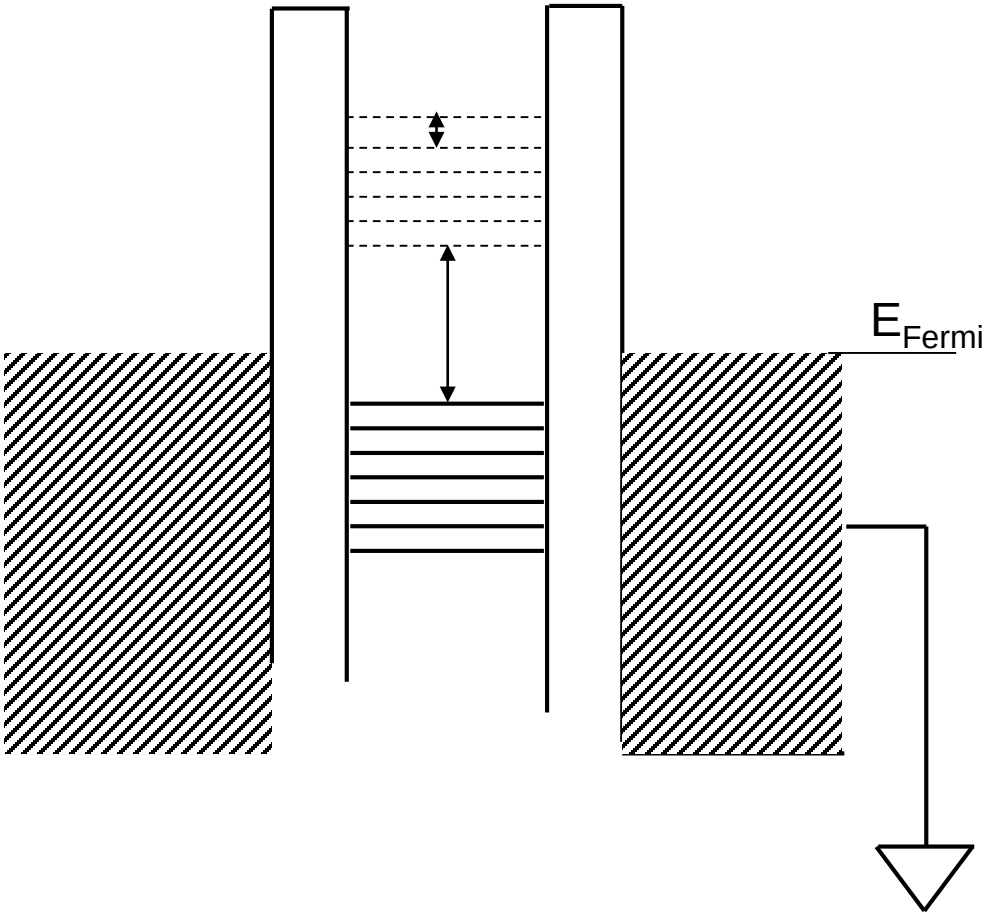


Band diagram with Coulomb "gap" and accounting for 0d states:

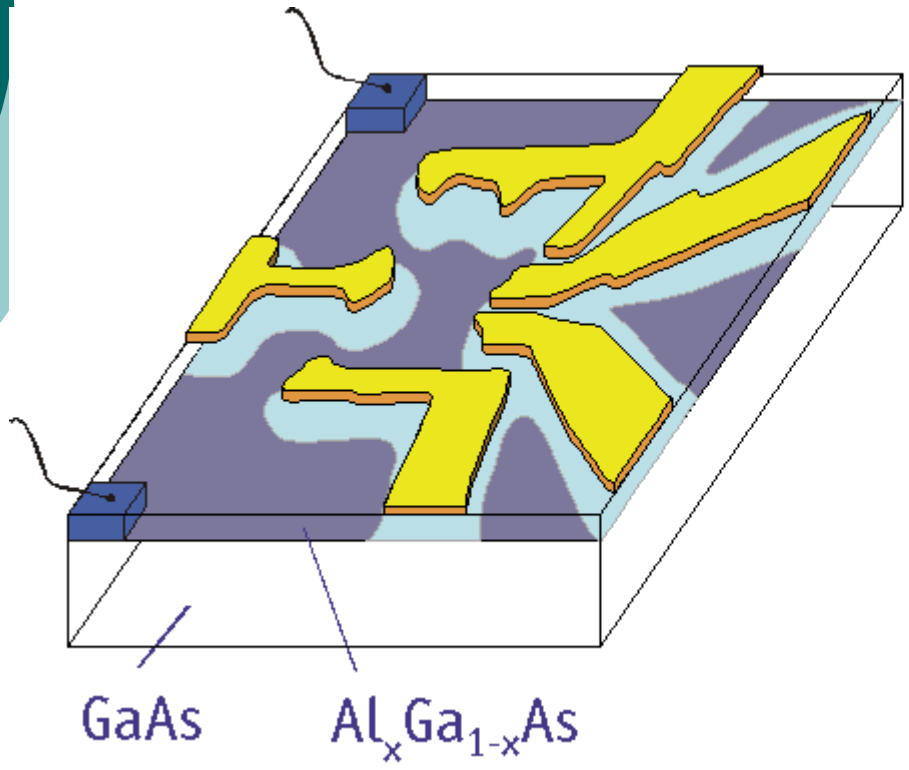


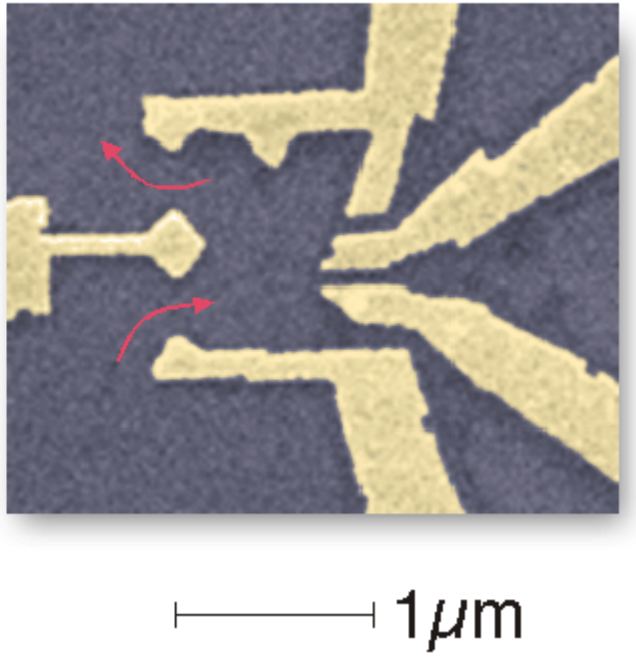


Band diagram with Coulomb "gap" and accounting for 0d states:



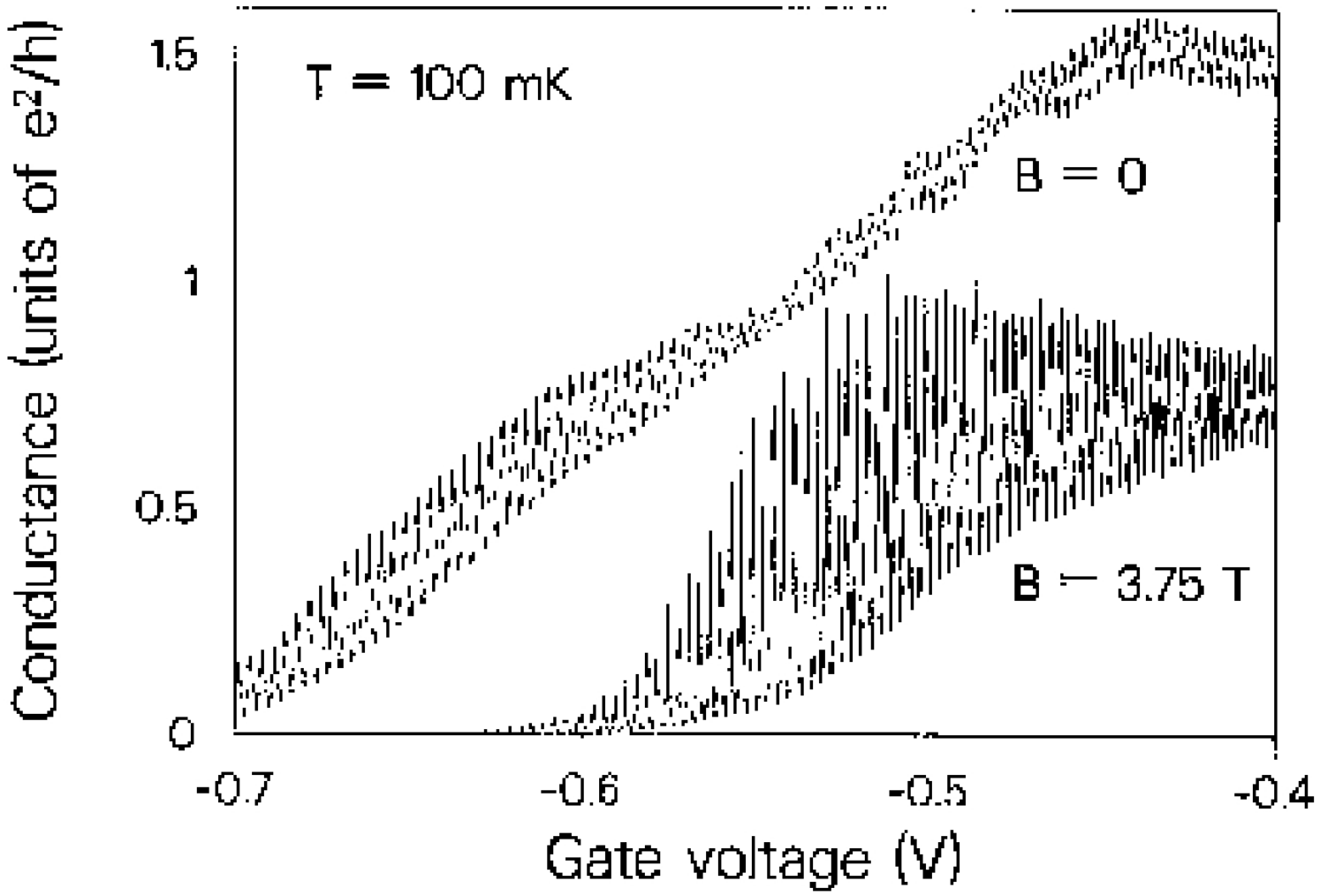
Experimental realization w/2DEGS





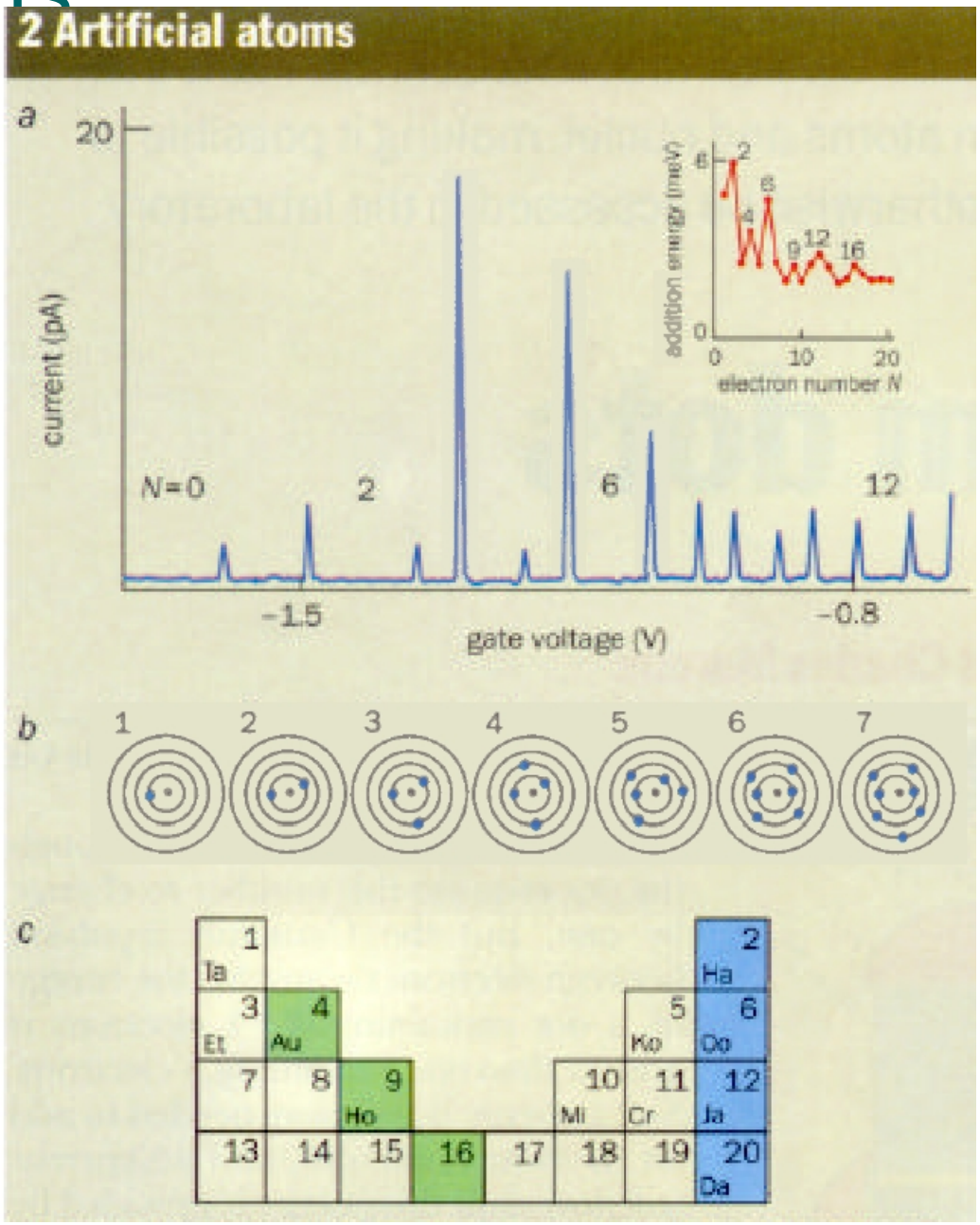
- http://marcuslab.harvard.edu/res.php

Results



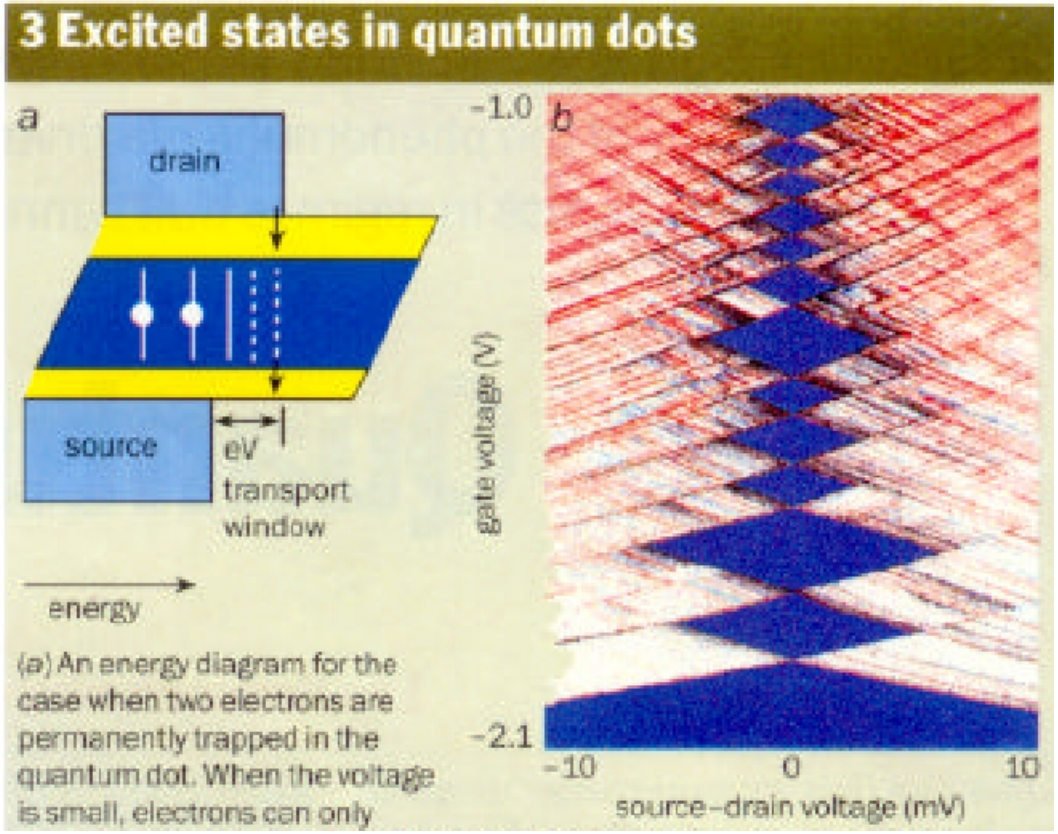
From .P. Kouwenhoven, C.M. Marcus, P.L. McEuen, S. Tarucha, R.M. Westervelt, and N.S. Wingreen
Electron Transport in Quantum Dots
Nato ASI conference proceedings, ed. By L. P. Kouwenhoven, G. Schön, L.L. Sohn (Kluwer, Dordrecht, 1997

Artificial atoms



Leo Kouwenhoven and Charles Marcus
Quantum Dots
Physics World, June 1998

Coulomb diamonds



Leo Kouwenhoven and Charles Marcus
Quantum Dots
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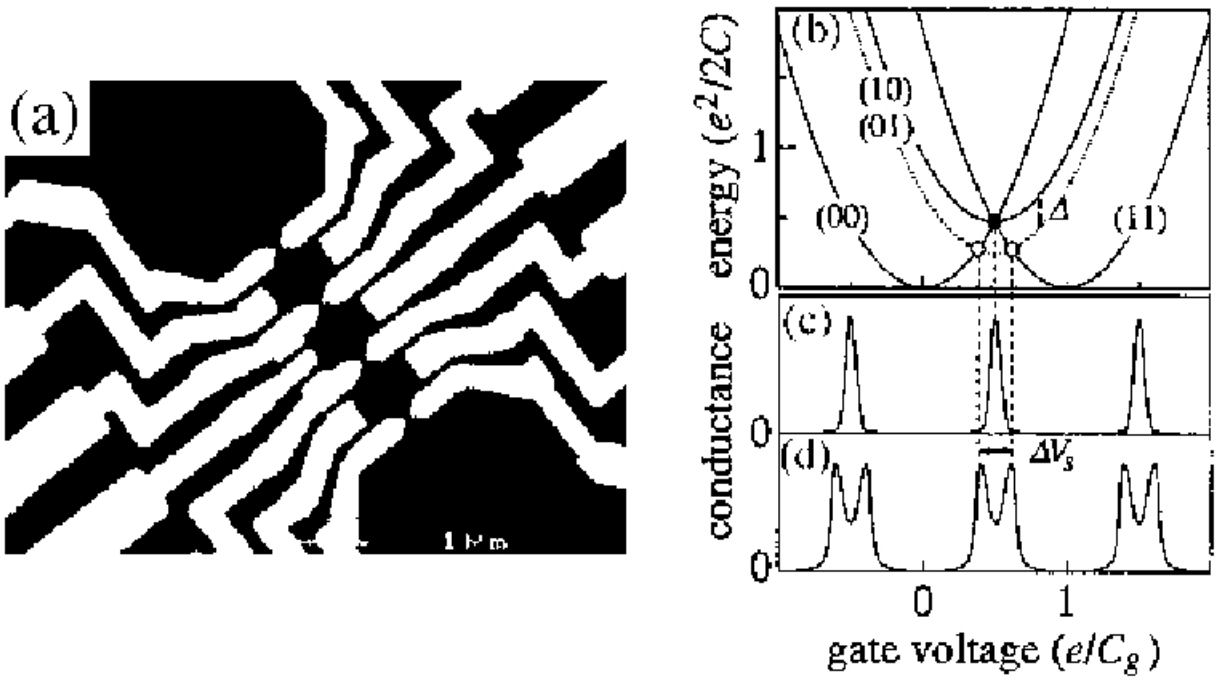
Artificial molecules



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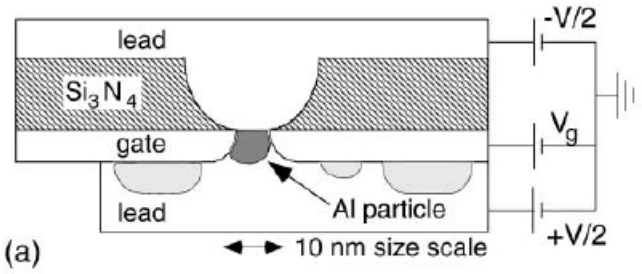
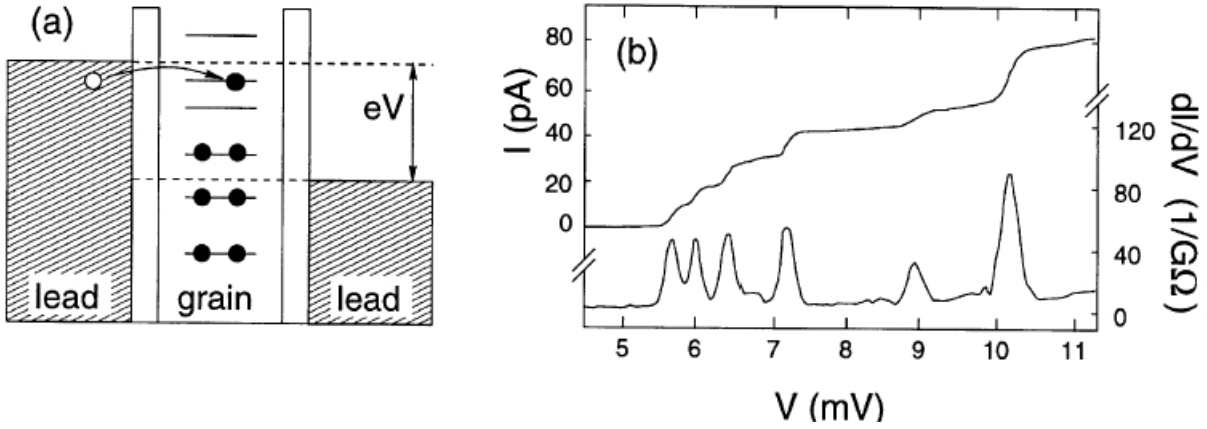
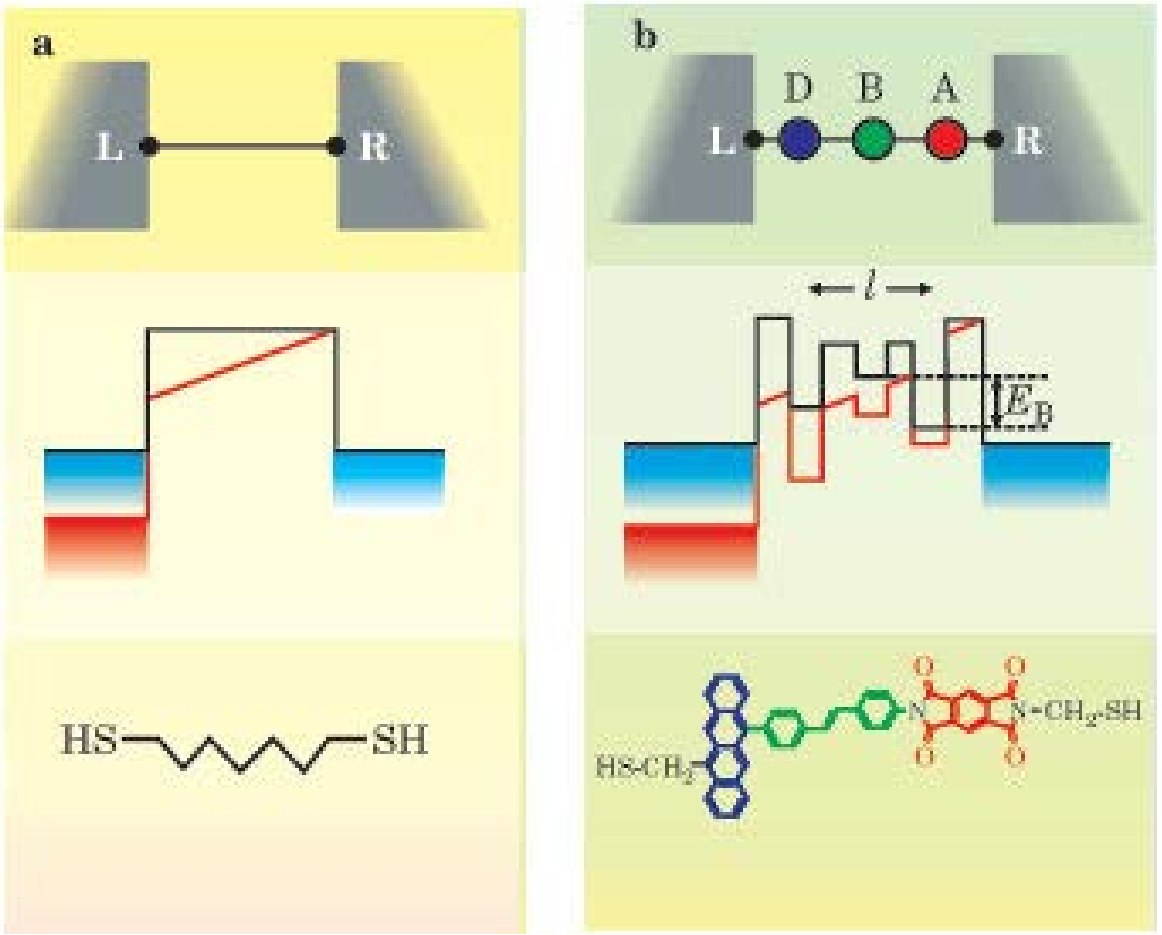
Metallic quantum dots





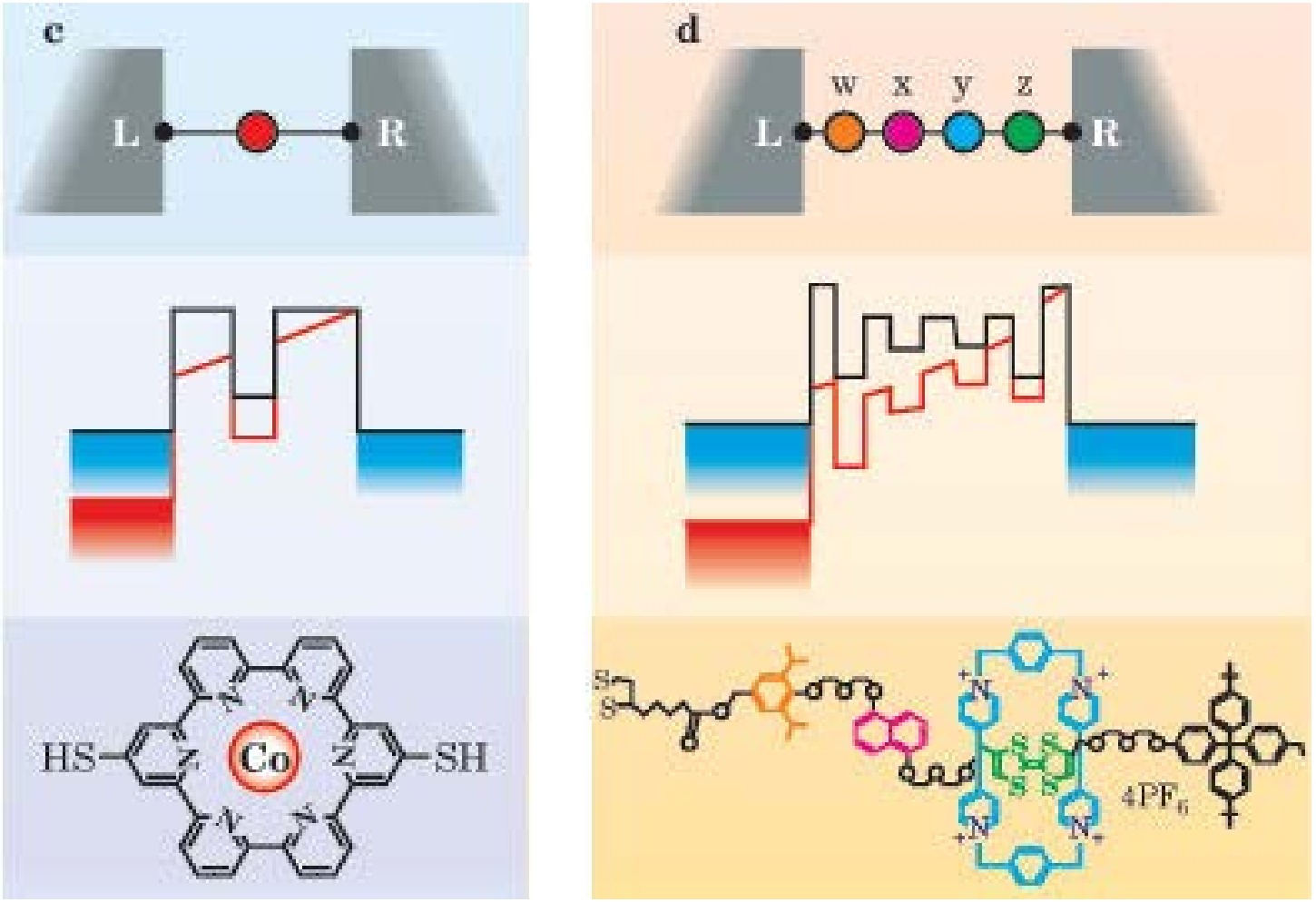
Spectroscopy of Discrete Energy Levels in Ultrasmall Metallic Grains, Jan von Delft and D. C. Ralph, *Physics Reports* **345**, 61 (2001).

Molecular electronics



J. Heath, Physics Today, May 2003

Molecular electronics



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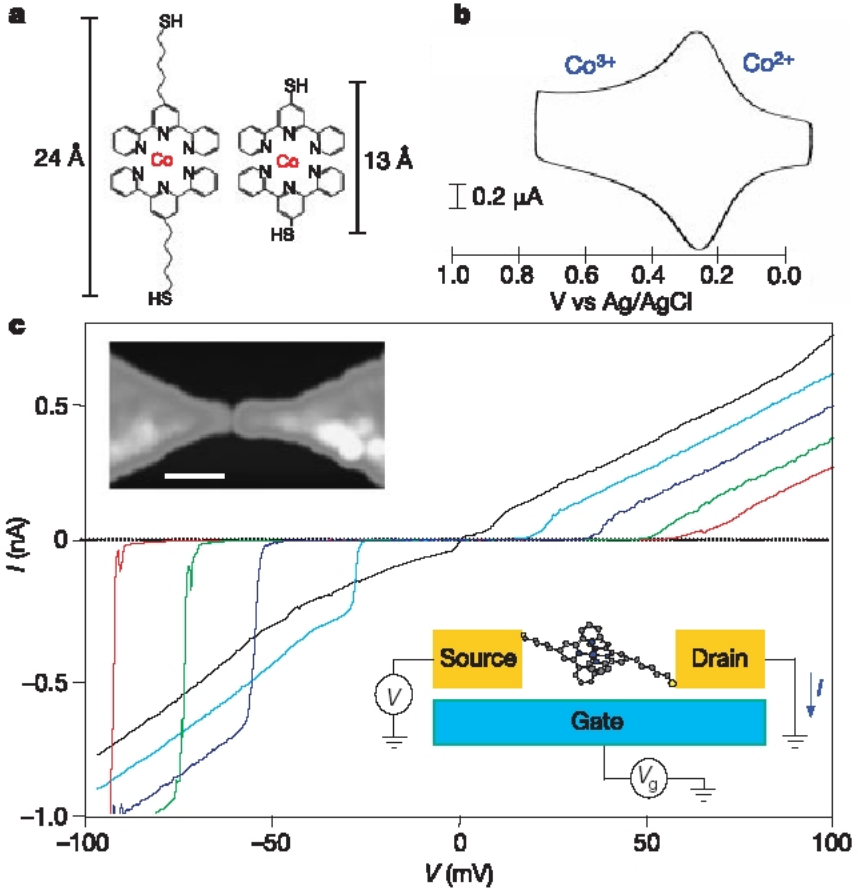
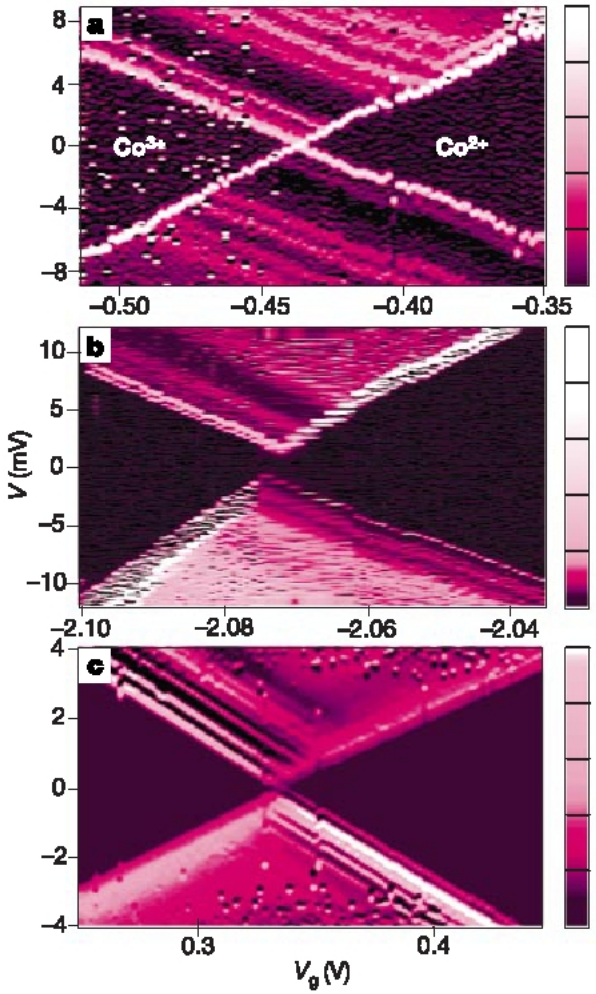


Figure 1 The molecules used in this study and their electronic properties. **a**, Structure of $[Co(tpy\text{-}(CH_2)_5\text{-}SH)_2]^{2+}$ (where tpy-$(CH_2)_5$-SH is 4′-(5-mercaptopentyl)-2,2′:6′,2″-terpyridinyl) and $[Co(tpy\text{-}SH)_2]^{2+}$ (where tpy-SH is 4′-(mercapto)-2,2′:6′,2″-terpyridinyl). The scale bars show the lengths of the molecules as calculated by energy minimization.



"Coulomb blockade and the Kondo effect in single-atom transistors," Jiwoong Park, Abhay N. Pasupathy, Jonas I. Goldsmith, Connie Chang, Yuval Yaish, Jason R. Petta, Marie Rinkoski, James P. Sethna, Hector D. Abruna, Paul L. McEuen & Daniel C. Ralph, Nature, 417, 722-725 (2002).